Name:_____ **Date:**_____ **Warm up 89**
Chemistry

Describe electronegativity.

the measure of an element's ability to attract electrons

Describe ionization energy.

the minimum amount of energy required to remove an electron

Write the following elements in order of INCREASING electronegativity. **Na, O, W, Cl, Ne, S**

Ne, Na, W, S, Cl, O

Write the following elements in order of increasing first ionization energy. **Zn, As, F, P, Sb, Li**

Li, Zn, Sb, As, P, F

Write the following elements in order of INCREASING atomic radii. **Ca, Na, Al, As, S, H**

H, S, Al, As, Na, Ca